PREPARATION AND APPLICATION OF A CHIRAL C₃-BUILDING BLOCK FOR AMINO ALCOHOL SYNTHESIS BY BAKERS' YEAST REDUCTION OF 1-ACYLOXY-3-AZIDO-2-PROPANONE

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Summary: Bakers' yeast-mediated reduction of $1-acyloxy-3-azido-2-propanone gives (S)-1-acyloxy-3-azido-2-propanol with high enantiomeric excess, which is a useful chiral C₃-building block for amino alcohols such as <math>\beta$ -blockers.

Recently trifunctional chiral building blocks with different hetero atom functionalities at both terminal carbons have been prepared from sulfurfunctionalized ketones by bakers' yeast-mediated asymmetric reduction.¹ The terminal hetero atom functional groups are very effective both for stereocontrol in the enantioselective reduction and for further manipulation of reduction products to natural product synthesis. In this paper, we wish to present a new trifunctional chiral building block for amino alcohol, (S)-1acyloxy-3-azido-2-propanol (2), prepared by the bakers' yeast-mediated reduction of 1-acyloxy-3-azido-2-propanone.

The substrates 1 were easily prepared by the reaction of 1-acyloxy-3-chloro-2-propanone with sodium azide² starting from 1,3-dichloro-2-propanone³ or 3-chloro-1,2-propanediol.⁴ A mixture of 40 g of pressed bakers' yeast (Oriental Yeast Co.) and 48 g of saccharose in 400 ml of water was stirred at room temperature for 30 min, then 10 ml of an ethanol solution of ketone 1 (8.0 mmol) was added to the fermenting yeast. Reaction progress was monitored by TLC until 1 was totally consumed. After usual work-up, the reduction product 2 was isolated by silica-gel column chromatography (hexane-ethyl acetate). Optical purity of 2 was determined by ¹H-NMR measurement or HPLC analysis of the corresponding (+)-MTPA ester. The configuration of 2a was defined to be S by comparison of the rotation of 3-azido-1,2-propanediol(2d) derived by hydrolysis of 2a, $[\alpha]_D^{23} -12.6^\circ$ (c 0.95, CH₃OH), with that of authentic sample, $[\alpha]_D^{23} -7.4^\circ$ (c 1.01, CH₃OH), prepared from (R)-3-tosyloxy-1,2-0-isopropylidene-1,2-propanediol, $[\alpha]_D^{23} -3.8^\circ$ (c 1.01, C₅H₅N), lit.⁵ $[\alpha]_D -7.3^\circ$ (c



3702

Ketone		- 4+		
	Time/h	Yield/%	%ee ^a	$[\alpha]_{\rm D}^{23}/^{\circ}$ (CH ₃ OH)
1a	0.5	70	>96	-12.6 (c 0.95)
	20	74 ^b	75 ^{b,c}	-10.8 (c 0.98) ^b
1b	б	95	78	-9.8 (c 0.98)
1c	14	74	90	-15.2 (c 1.25)

Table 1. The bakers' yeast reduction of ketone 1 to alcohol 2

^a Determined by ¹H NMR of the corresponding (+)-MTPA ester. ^b Values for (S)-3-azido-1,2propanediol(2d). ^C Determined by HPLC of the corresponding di-MTPA ester.

5.5, C_5H_5N). The (S)-configuration of 2b and 2c were supported by its negative sign of the rotation in agreement with that of authentic samples prepared by acylation of 2d. The results of the bakers' yeast reduction are listed in Table 1.

As shown in Table 1, the acetyl group is the best choice of protection of the hydroxyl group in ketone 1. The reduction of 1a proceeds much faster than that of 1b and 1c, and the highest enantioselectivity can be achieved by the reduction of 1a. However, a longer reduction time in the case of 1a resulted in formation of 2d. The low enantiomeric excess of 2d may be due to kinetic resolution in hydrolysis of 2a in favor of formation of (R)-2d and/or to reduction of 1d, produced first by hydrolysis of 1a, to give (R)-2d.6

The utility of 2a as a chiral synthon involving nitrogen atom was demonstrated in the synthesis of 3d, a precursor of β -blockers⁷ as follows: phenoxycarbonylation of 2a (ClCO₂Ph-DMAP, 83%) and reduction of azide (H₂/10% Pd-C) gave oxazolidinone 3a in 87% yield, $[\alpha]_D^{23}$ +47.6° (c 0.93, C₂H₅OH). Deacetylation of 3a with K_2CO_3 in C_2H_5 OH gave (S)-5-hydroxymethyl-2oxazolidinone 3d in 97% yield, [α]²³_D +29.1° (c 2.7, C₂H₅OH), lit.⁷ [α]_D +29.7° (c 2.7, C₂H₅OH).

Thus, the bakers' yeast reduction of 1-acetoxy-3-azido-2-propanone provides a useful chiral building block for the synthesis of amino alcohols.

References

- T. Fujisawa, T. Itoh, M. Nakai, and T. Sato, Tetrahedron Lett., <u>26</u>, 771 (1985); T. Itoh, A. Yoshinaka, T. Sato, and T. Fijisawa, Chem. Lett., <u>1985</u>, 1679; T. Fujisawa, E. Kojima, T. Itoh, and T. Sato, *ibid.*, <u>1985</u>, 1751; T. Sato, Y. Okumura, J. Itai, and T. Fujisawa, *ibid.*, <u>1988</u>, 1537; R. Tanikaga, K. Hosoya, and K. Kaji, J. Chem. Soc., Chem. Commun., <u>1987</u>, 389.
 F. C. Hartman, *Biochemistry*, 9, 1776 (1970).
- 3. E. R. Clark and J. G. B. Howes, J. Chem. Soc., 1956, 1152.
- 4. N. Bischofberger, H. Waldmann, T. Saito, E. S. Simon, W. Lees, M. D.
- Bednarski, and G. M. Whitesides, J. Org. Chem., <u>53</u>, 3457 (1988). 5. J. C. Sowden and H. O. L. Fischer, J. Am. Chem. Soc., <u>64</u>, 1291 (1942).
- 6. Bakers' yeast reduction of α -hydroxyacetophenone and α -acetoxyacetophenone is known to afford the reduction products with opposite absolute configuration, respectively. A. Monzocchi, A. Fiecchi, and E. Santaniello, J. Org. Chem., 53, 4405 (1988).
- 7. For Éxample, G. Cardillo, N. Orena, S. Sandri, and C. Tomasini, Tetrahedron, 43, 2505 (1987).

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